Total No. of Questions: 11

[ Total No. of Printed Pages : 4

## CHEMSEM-102

## M.Sc. (Ist Semester) Examination Dec., 2022 CHEMISTRY

Paper - CC-2

(Organic Chemistry-I)

Time: 3 Hours ] [ Maximum Marks: 40

Section-A (Marks :  $1 \times 10 = 10$ )

Note: Answer all ten questions (Answer limit 50 words). Each question carries1 mark.

Section–B (Marks :  $3 \times 5 = 15$ )

Note: Answer all five questions. Each question has internal choice (Answer limit200 words). Each question carries 3 marks.

Section–C (Marks:  $5 \times 3 = 15$ )

**Note**: Answer any *three* questions out of five (Answer limit **500** words). Each question carries **5** marks.

## Section-A

- 1. (i) Crown ether
  - (ii) Cryptands

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|                           | (iii)  | Cyclodextrins   |  |  |
|---------------------------|--------|---|--|--|
|                           | (iv)   | Isotopic effect   |  |  |
|                           | (v)    | Steric effect   |  |  |
|                           | (vi)   | Nitrene   |  |  |
|                           | (vii)  | Taft equation   |  |  |
|                           | (viii) | Ortho/para ratio  |  |  |
|                           | (ix)   | Annulenes   |  |  |
|                           | (x)    | Regioselectivity  |  |  |
|                           |        | Section-B   |  |  |
| 2. Explain the following: |        |   |  |  |
|                           | (a)    | Cyclopentadienyl cation is an antiaromatic compound whereas                       |  |  |
|                           |        | cyclopropenyl cation is an aromatic.  |  |  |
|                           | (b)    | Allyl and benzyl halides are more reactive than alkyl halides towards             |  |  |
|                           |        | nucleophilic substitution reactions.  |  |  |
|                           |        | Or  |  |  |
|                           | (a)    | Ethyl acetoacetate has about 10% enol content in ethanole solution which          |  |  |
|                           |        | increase to 46% when dissolved in hexane. Explain.                                |  |  |
|                           | (b)    | What is Huckel Rule for Aromaticity ? Gives its limitations ?                     |  |  |
| 3.                        | Write  | short notes on the following:   |  |  |
|                           | (i)    | Hammonds Postulates   |  |  |
|                           | (ii)   | Curtin Hammett Principle  |  |  |
|                           |        | Or  |  |  |
|                           | Descr  | ibe any two methods of determining of reaction mechanism.                         |  |  |
| BF                        | RI-2   | ( 2 ) <b>CHEMSEM-102</b>  |  |  |
|                           | 3.     | (iv) (v) (vi) (vii) (viii) (ix) (x)  2. Expla (a) (b)  (a) (b)  3. Write (i) (ii) |  |  |

| 5. | Predict the product and sketch a suitable mechanism for the following reactions: |   |  |  |  |  |  |
|----|--|---|--|--|--|--|--|
|    | (i)  | $ + HCONR2 + POCl3 \longrightarrow ? $              |  |  |  |  |  |
|    | (ii)   | $ + CO + HC1 \xrightarrow{AlCl_3} ? $               |  |  |  |  |  |
|    |  | Or  |  |  |  |  |  |
|    | Explain the following:   |   |  |  |  |  |  |
|    | (i)  | Hydroboration                                       |  |  |  |  |  |
|    | (ii)   | Michael Reaction                                    |  |  |  |  |  |
| 6. | Discuss the $E_1$ and $E_2$ mechanism.   |   |  |  |  |  |  |
|    | Or   |   |  |  |  |  |  |
|    | Describe the Stereochemical preference in elimination reaction.                  |   |  |  |  |  |  |
|    |  | Section-C   |  |  |  |  |  |
| 7. | Briefly describe the following:  |   |  |  |  |  |  |
|    | (i)  | Aromaticity in Benzanoid and Non-benzanoid compound |  |  |  |  |  |
|    | (ii)   | Psi aromaticity                                     |  |  |  |  |  |
| 8. | . Explain the following :  |   |  |  |  |  |  |
|    | (i)  | Stability and reactivity of carbene                 |  |  |  |  |  |
|    | (ii)   | Kinetics and thermodynamics control of reaction     |  |  |  |  |  |
| RI | RI-2   | CHEMSEM-102 PTO                                     |  |  |  |  |  |

Describe  $SE^1$  and  $SE^2$  mechanism with suitable example.

Discuss the IPSO attack with suitable example.

Or

| 9.         | Explain | the  | following  | • |
|------------|---------|------|------------|---|
| <i>-</i> . | LApium  | tric | TOHO WILLS | • |

- (i) Effect of leaving group on the reactivity
- (ii) Effect of solvent polarity on the reactivity
- 10. Predict the structure of alkenes expected from dehydrohelogenation of the following alkyl halides :
  - (i) 2-chlorobutane
  - (ii) 2-chloro-3-methyl butane
  - (iii) 3-chloro-2-methyl butane
  - (iv) 2-chloro-3-methyl pentane
  - (v) 2-chloro-3, 3-dimethyl pentane

## 11. Discuss the following:

- (i) Diazonium coupling
- (ii) Asymmetric epoxidation